# Enantiospecific Synthesis of the (4R)-1-Azabicyclo[2.2.1]heptane Ring System 

Peter G. Houghton, Guy R. Humphrey, Derek J. Kennedy, D. Craig Roberts and Stanley H. B. Wright<br>Merck Sharp and Dohme Research Laboratories, Hertford Road, Hoddesdon, Herts. EN11 9BU, UK


#### Abstract

An enantioselective synthesis of (4R)-1-azabicyclo[2.2.1]heptane derivatives is described commencing from readily available trans-4-hydroxy-L-proline which is converted into the key intermediate (3R)-N-(tert-butoxycarbonyl)-3-methylsulfonyloxypyrrolidine 4. Reaction of the sulfonate ester 4 with an enolate anion yields a mixture of ( $3 R$ )-pyrrolidinylacetic esters 8 and 9 which are reduced to the corresponding alcohols 10 and 11. Conversion of the alcohols into the sulfonate esters 12 and 13 followed by deprotection of the pyrrolidine nitrogen leads to cyclisation yielding the (4R)-1azabicyclo[2.2.1] heptane derivatives 14 and 15.


Heterocyclic derivatives of 1-azabicyclo[2.2.1]heptane are potent muscarinic agonists and have potential for the treatment of senile dementia of the Alzheimer type. ${ }^{1,2}$ Investigation of a number of agonists identified ( $3 R, 4 R$ )-3-(3-cyclopropyl-1,2,4-oxadiazol-5-yl)-1-azabicyclo[2.2.1]heptane as a new drug candidate, ${ }^{3}$ and thus larger quantities of material for further biological evaluation were required. The large scale preparation of chiral intermediates for this type of heterocycle has been reported in an earlier communication, ${ }^{4}$ which described a novel route to 1 -azabicyclo[2.2.1]heptane-3-carboxylic acid ester enantiomers involving resolution assisted by a chiral auxiliary. However, unless the unwanted isomer can be epimerised, the resolution of diastereoisomers is not a very efficient method for the preparation of a chiral compound. We now report an enantiospecific synthesis of the oxadiazole derivative taking advantage of a chiral pool starting material to introduce the $4 R$ asymmetric bridgehead carbon atom of the 1-azabicyclo[2.2.1]heptane ring system.

## Results and Discussion

As in our earlier method, the azabicyclic ring system is constructed by forming a two-carbon bridge across a pyrrolidine ring onto the nitrogen atom. However in this new method the stereochemistry at the bridgehead is established by stereospecifically introducing the side chain by displacement, with inversion, of methanesulfonate ion from a chiral pyrrolidinyl alcohol derivative. X-ray crystallography ${ }^{3}$ had established the absolute configuration of the exo-isomer 14, and thus the key intermediate for this approach is $(3 R)-(-)-3-$ hydroxypyrrolidine obtained by decarboxylation of readily available trans-4-hydroxy-L-proline 1 by the method of Hashimoto. ${ }^{5}$ In our hands we found isolation of the pyrrolidine as the crystalline hydrogen maleate salt 2 ( $74 \%$ yield) was more practicable than the hygroscopic hydrochloride, which was difficult to handle on the large scale. Protection of the pyrrolidine as the tert-butoxycarbonyl derivative 3 and conversion into the methanesulfonate ester 4 proceeded in $98 \%$ overall yield (Scheme 1).

Reaction of cyclopropanecarboxamide oxime 5 with an excess of diethyl malonate gave a mixture containing the oxadiazole acetic ester 6 with a little of the bis by-product 7 (Scheme 2). Distillation gave recovered diethyl malonate and then the product 6 in $69 \%$ yield. Generation of the enolate of the ester 6 by treatment with sodium hydride or potassium tertbutoxide in dimethylformamide (DMF) followed by heating with the methanesulfonate 4 at $100^{\circ} \mathrm{C}$ gave the alkylated products 8 and 9 , but with up to $23 \%$ of the product formed by


Scheme 1 Reagents: i, cyclohexanol, 2-cyclohexen-1-one; ii, maleic acid, $\mathrm{Me}_{2} \mathrm{CHOH}$; iii, $\left(\mathrm{Bu}^{\mathrm{t} O C O}\right)_{2} \mathrm{O}, \mathrm{NaHCO}_{3}, \mathrm{H}_{2} \mathrm{O}$; iv, $\mathrm{MeSO}_{2} \mathrm{Cl}$, $\mathrm{NEt}_{3}, \mathrm{EtOAc}$
elimination of methanesulfonic acid from the pyrrolidine sulfonate. Attempts to use the organic bases diisopropylethylamine or tributylamine were unsuccessful as no reaction occurred on heating the reactants with the amines in 1,3-dimethyl-3,4,5,6-tetrahydropyrimidin-2( $1 H$ )-one (DMPU) at $80^{\circ} \mathrm{C}$. However, heating the reactants in DMPU with $1,8-$ diazabicyclo[5.4.0]undec-7-ene (DBU) at $80^{\circ} \mathrm{C}$ gave the products 8 and 9 with $<15 \%$ of the eliminated product. Better results were obtained by stirring the methanesulfonate 4 and DBU with an excess of ester 6 at $55^{\circ} \mathrm{C}$ without solvent to give the alkylated products 8 and 9 , containing only $6 \%$ of the eliminated product, as an oil in $97 \%$ crude yield. Some diastereoselection occurred in this reaction generating another chiral centre, as the two isomers 8 and 9 were obtained in the ratio $1.6: 1$. A sample was purified by column chromatography for analysis, but the bulk of the crude ester was reduced to the alcohols 10 and 11. The ester was recovered unchanged after attempted reduction with sodium bis(2-methoxyethoxy)aluminium hydride, and reduction of the ester with diisobutylaluminium hydride gave the corresponding aldehyde. Reduction of the ester with sodium borohydride activated by methanol, ${ }^{6}$ a method well suited for large scale work, gave the alcohols 10 and 11 in $85 \%$ yield. GC/MS analysis showed the product to be $95 \%$ pure with diastereoisomers 10 and 11 in a ratio of $1.9: 1$. A sample of the pure $\left(2 R, 3^{\prime} S\right)$ enantiomer 10 was isolated by


6



8


7


9 iii, iv





15

Scheme 2 Reagents: i, $\mathrm{CH}_{2}\left(\mathrm{CO}_{2} \mathrm{Et}\right)_{2}$, toluene; ii, 4, DBU; iii, $\mathrm{NaBH}_{4}$, THF, MeOH ; iv, $\mathrm{MeSO}_{2} \mathrm{Cl}, \mathrm{NEt}_{3}$, EtOAc ; v, $\mathrm{CF}_{3} \mathrm{CO}_{2} \mathrm{H}, \mathrm{Me}_{3} \mathrm{COH}$; vi, $\mathrm{Na}_{2} \mathrm{CO}_{3}, \mathrm{H}_{2} \mathrm{O}, \mathrm{Me}_{3} \mathrm{COH}$; vii, KOBu', EtOH
column chromatography for analysis. The crude alcohols 10 and 11 were converted into the methanesulfonate esters 12 and 13 in quantitative yield, and an analytical sample of the $\left(2 R, 3^{\prime} S\right)$ enantiomer 12 obtained by chromatography.

Deprotection of the pyrrolidine under acid conditions followed by neutralisation of the solution gave a mixture of the bicyclic products 14 and 15 in a ratio of $2: 1$ in $69 \%$ yield. Separation of the desired exo-isomer 14 may be achieved by chromatography, or more conveniently for large scale work by fractional crystallisation of the toluene-p-sulfonate salts. The pure ( $3 R, 4 R$ )-azabicyclo[2.2.1] heptane 14 was obtained in $27 \%$ yield, and a sample of the pure endo-isomer 15 isolated by chromatography for identification. The yield of the exo-isomer 14 was increased further by epimerisation ${ }^{3}$ of the endo-isomer 15 followed by crystallisation of the toluene-p-sulfonate salt from the thermodynamic equilibrium mixture of $3: 1$ exo :endo isomers.

The preparation of the oxadiazole derivatives 14 and 15 demonstrates this novel process which clearly has wider application in the synthesis of other ( $4 R$ )-azabicyclo[2.2.1]heptanes.

## Experimental

M.p.s were determined on a Büchi 510 apparatus and are uncorrected. ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR spectra were obtained on a Bruker AM- 250 spectrometer with tetramethylsilane as internal standard. All $J$ values are in Hz . GC analyses were carried out with a Hewlett Packard 5790 gas chromatograph with a cool on-column injector (Column CPSIL5CB, $75^{\circ} \mathrm{C}$ for 2 min and then $10^{\circ} \mathrm{C} \mathrm{min}^{-1}$ to $300^{\circ} \mathrm{C}$ ). GC/MS analyses were carried out with a Finnegan Mat TSQ70 spectrometer. $[\alpha]_{\mathrm{D}}$ Values are given in units of $10^{-1} \mathrm{deg} \mathrm{cm}^{2} \mathrm{~g}^{-1}$.
(3R)-3-Hydroxypyrrolidinium Hydrogen Maleate 2.$(2 S, 4 R)-(-)$-4-Hydroxypyrrolidine-2-carboxylic acid $(1.0 \mathrm{~kg}$, $7.6 \mathrm{~mol})$ suspended in cyclohexanol ( $5.0 \mathrm{dm}^{3}$ ) and cyclohex-2-en-1-one ( $100 \mathrm{~cm}^{3}$ ) was heated under reflux at $155^{\circ} \mathrm{C}$ for 5.5 h until all the solid had dissolved. The clear red solution was cooled to $25^{\circ} \mathrm{C}$ and maleic acid ( $885 \mathrm{~g}, 7.6 \mathrm{~mol}$ ) added over 30 $\min$ maintaining the temperature $<35^{\circ} \mathrm{C}$. On complete addition crystallisation occurred and the mixture was stirred at $25^{\circ} \mathrm{C}$ for 30 min . Ethyl acetate $\left(10 \mathrm{dm}^{3}\right)$ was added over 1 h and then the suspension stirred at $20^{\circ} \mathrm{C}$ for 2 h . The crystalline solid was collected, washed with a mixture of ethyl acetate ( $2 \mathrm{dm}^{3}$ ) and cyclohexanol $\left(1 \mathrm{dm}^{3}\right)$ and then ethyl acetate $\left(3 \mathrm{dm}^{3}\right)$ to give the pyrrolidine salt $2\left(1.14 \mathrm{~kg}, 74 \%\right.$ ), m.p. $90-91^{\circ} \mathrm{C}$ (Found: C , 47.25; $\mathrm{H}, 6.4 ; \mathrm{N}, 6.9 . \mathrm{C}_{4} \mathrm{H}_{9} \mathrm{NO} \cdot \mathrm{C}_{4} \mathrm{H}_{4} \mathrm{O}_{4}$ requires $\mathrm{C}, 47.3$; H , 6.45; N, 6.9\%); $[\alpha]_{\mathrm{D}}-5.4\left(c \mathrm{l}\right.$ in $\left.\mathrm{H}_{2} \mathrm{O}\right), \delta_{\mathrm{H}}\left(\mathrm{CD}_{3} \mathrm{OD}\right) 2.05(2 \mathrm{H}$, $\mathrm{m}, \mathrm{CCH}_{2} \mathrm{C}$ ), 3.2-3.6[4 H, m, $\left.\left(\mathrm{NCH}_{2}\right)_{2}\right], 4.57(1 \mathrm{H}, \mathrm{m}, \mathrm{OCH})$ and $6.29(2 \mathrm{H}, \mathrm{s}, \mathrm{CH}=\mathrm{CH})$.
(3R)-N-(tert-Butoxycarbonyl)-3-hydroxypyrrolidine 3.-A solution of the pyrrolidinium hydrogen maleate $2(3.6 \mathrm{~kg}, 17.7$ mol ) in water ( $10.8 \mathrm{dm}^{3}$ ) was added over 15 min to a suspension of sodium hydrogen carbonate $(7.42 \mathrm{~kg}, 88.3 \mathrm{~mol})$ in water ( 29 $\mathrm{dm}^{3}$ ) at $20^{\circ} \mathrm{C}$. Di-tert-butyl dicarbonate ( $4.64 \mathrm{~kg}, 21.3 \mathrm{~mol}$ ) was added to the solution and the mixture stirred for 65 h . Ethyl acetate ( $10 \mathrm{dm}^{3}$ ) was added and the mixture filtered. The aqueous layer was separated from the filtrate and extracted with ethyl acetate $\left(10 \mathrm{dm}^{3}\right)$. The combined extracts were dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$ and evaporated under reduced pressure to give the hydroxypyrrolidine 3 as an oil ( $3.3 \mathrm{~kg}, 100 \%$ ), which crystallised, m.p. $62-65^{\circ} \mathrm{C}$ (Found: $\mathrm{C}, 57.8 ; \mathrm{H}, 9.1 ; \mathrm{N}, 7.5$. $\mathrm{C}_{9} \mathrm{H}_{17} \mathrm{NO}_{3}$ requires $\mathrm{C}, 57.7 ; \mathrm{H}, 9.15 ; \mathrm{N}, 7.5 \%$ ); $[\alpha]_{\mathrm{D}}-22.7(c 1$ in $\left.\mathrm{CH}_{2} \mathrm{Cl}_{2}\right) ; \delta_{\mathrm{H}}\left[\left(\mathrm{CD}_{3}\right)_{2} \mathrm{SO}\right.$ at $\left.60^{\circ} \mathrm{C}\right] 1.40\left(9 \mathrm{H}, \mathrm{s}, \mathrm{CMe}_{3}\right), 1.73(1$ $\mathrm{H}, \mathrm{m}, 4-\mathrm{H}), 1.95(1 \mathrm{H}, \mathrm{dtd}, J 5.0,9.0$ and $11.0,4-\mathrm{H}), 3.11(1 \mathrm{H}$, ddd, $J 1.8,2.5$ and $11.0,2-\mathrm{H}), 3.3\left(3 \mathrm{H}, \mathrm{m}, 2-\mathrm{H}\right.$ and $\left.5-\mathrm{H}_{2}\right)$ and $4.21(1 \mathrm{H}, \mathrm{m}, 3-\mathrm{H})$; GC analysis $99 \% R_{\mathrm{t}} 7.9 \mathrm{~min}$.
(3R)-N-(tert-Butoxycarbonyl)-3-methylsulfonyloxypyrrolidine 4.-Methanesulfonyl chloride ( $1.68 \mathrm{dm}^{3}, 21.7 \mathrm{~mol}$ ) was added over 1 h to a solution of the alcohol $3(3.39 \mathrm{~kg}, 18.1 \mathrm{~mol})$ and triethylamine $\left(5.1 \mathrm{dm}^{3}, 36.5 \mathrm{~mol}\right)$ in ethyl acetate $\left(50 \mathrm{dm}^{3}\right)$ maintained at $-5^{\circ} \mathrm{C}$. The mixture was stirred at $-5^{\circ} \mathrm{C}$ for 30 min and then quenched with water ( $20 \mathrm{dm}^{3}$ ) added over 10 min . The organic phase was separated and washed sequentially with hydrochloric acid ( $10 \mathrm{dm}^{3}, 1 \mathrm{~mol} \mathrm{dm}{ }^{-3}$ ) and saturated sodium hydrogen carbonate solution ( $10 \mathrm{dm}^{3}$ ). The solution was dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$ and evaporated to give the methanesulfonate 4 as an oil ( $4.725 \mathrm{~kg}, 98 \%$ ) (Found: C, $45.1 ; \mathrm{H}, 7.2 ; \mathrm{N}, 5.2 ; \mathrm{S}, 12.0$. $\mathrm{C}_{10} \mathrm{H}_{19} \mathrm{NO}_{5} \mathrm{~S}$ requires $\left.\mathrm{C}, 45.25 ; \mathrm{H}, 7.2 ; \mathrm{N}, 5.3, \mathrm{~S}, 12.1 \%\right) ;[\alpha]_{\mathrm{D}}$ $-27\left(c 1\right.$ in $\left.\mathrm{CH}_{2} \mathrm{Cl}_{2}\right) ; \delta_{\mathrm{H}}\left[\left(\mathrm{CD}_{3}\right)_{2} \mathrm{SO}\right.$ at $\left.60^{\circ} \mathrm{C}\right] 2.1(2 \mathrm{H}, \mathrm{m}, 4-$ $\left.\mathrm{H}_{2}\right), 3.20\left(3 \mathrm{H}, \mathrm{s}, \mathrm{SO}_{2} \mathrm{CH}_{3}\right), 3.25(1 \mathrm{H}, \mathrm{m}, 5-\mathrm{H}), 3.4(1 \mathrm{H}, \mathrm{m}, 5-$ $\mathrm{H}), 3.45(1 \mathrm{H}$, dd, $J 3.4$ and $12.5,2-\mathrm{H}), 3.55(1 \mathrm{H}$, dd, $J 4.0$ and $12.5,2-\mathrm{H})$ and $5.24(1 \mathrm{H}, \mathrm{m}, 3-\mathrm{H}) ; \mathrm{GC}$ analysis $99 \% R_{\mathrm{t}} 3.1 \mathrm{~min}$.

Ethyl 2-(3-Cyclopropyl-1,2,4-oxadiazol-5-yl)acetate 6.-A solution of cyclopropanecarboxamide oxime ${ }^{7}$ ( $5.39 \mathrm{~kg}, 53.9$ mol ) and diethyl malonate ( $24.5 \mathrm{~kg}, 160 \mathrm{~mol}$ ) in toluene ( $54 \mathrm{dm}^{3}$ ) was heated under reflux for 21 h . The aqueous ethanol layer
which separated from the condensate was periodically removed. GC analysis of reaction mixture $64 \% R_{\mathrm{t}} 2.4 \mathrm{~min}$ (diethyl malonate), $33 \% R_{\mathrm{t}} 7.6 \mathrm{~min}$ (oxadiazole 6 ), $3 \% R_{\mathrm{t}} 12.3 \mathrm{~min}$ (byproduct 7). The reaction mixture was cooled to $20^{\circ} \mathrm{C}$, washed with brine ( $3 \times 5 \mathrm{dm}^{3}$ ) and evaporated under reduced pressure. The residue was distilled through a short fractionating column to give recovered diethyl malonate ( 19.9 kg ), b.p. $60-70^{\circ} \mathrm{C} / 2$ mbar* and oxadiazole 6 as an oil $(7.26 \mathrm{~kg}, 69 \%$ ), b.p. $90-$ $100^{\circ} \mathrm{C} / 1 \mathrm{mbar}$ (Found: $\mathrm{C}, 54.85 ; \mathrm{H}, 6.2 ; \mathrm{N}, 14.0 . \mathrm{C}_{9} \mathrm{H}_{10} \mathrm{~N}_{2} \mathrm{O}_{3}$ requires $\mathrm{C}, 55.1 ; \mathrm{H}, 6.15 ; \mathrm{N}, 14.3 \%) ; \delta_{\mathrm{H}}\left(\mathrm{CD}_{2} \mathrm{Cl}_{2}\right) 1.0(4 \mathrm{H}, \mathrm{m}$, cyclopropyl, $\mathrm{CH}_{2} \mathrm{CH}_{2}$ ), $1.28(3 \mathrm{H}, \mathrm{t}, J 7.0, \mathrm{Me}), 2.09(1 \mathrm{H}, \mathrm{m}$, cyclopropyl CH$), 3.93\left(2 \mathrm{H}, \mathrm{s}, \mathrm{CH}_{2} \mathrm{CO}\right)$ and $4.20(2 \mathrm{H}, \mathrm{q}, J 7.0$, $\mathrm{CH}_{2}$ ); GC analysis $98 \% R_{\mathrm{t}} 7.6 \mathrm{~min}$.

Ethyl (2R,3'R)- and (2S,3'R)-2-(3-Cyclopropyl-1,2,4-oxa-diazol-5-yl)-2-(N-tert-butoxycarbonylpyrrolidin-3-yl)acetates 8 and 9.-Diazabicyclo[5.4.0]undec-7-ene ( $4.6 \mathrm{~kg}, 30 \mathrm{~mol}$ ) was added over 20 min to a mixture of mesylate $4(4.22 \mathrm{~kg}, 15.9 \mathrm{~mol})$ and oxadiazole $6(6.24 \mathrm{~kg}, 31.8 \mathrm{~mol})$. The temperature increased from $20^{\circ} \mathrm{C}$ to $45^{\circ} \mathrm{C}$ during this time, and then the mixture was heated at $54^{\circ} \mathrm{C}$ for 30 h . The mixture was cooled to $20^{\circ} \mathrm{C}$ and partitioned between ethyl acetate ( $12.5 \mathrm{dm}^{3}$ ) and hydrochloric acid $\left(8 \mathrm{dm}^{3}, 1 \mathrm{~mol} \mathrm{dm}^{-3}\right)$. The organic phase was washed sequentially with hydrochloric acid $\left(4 \mathrm{dm}^{3}, 1 \mathrm{~mol} \mathrm{dm}^{-3}\right)$ then brine $\left(2 \times 5 \mathrm{dm}^{3}\right)$ and dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$. The solution was evaporated under reduced pressure to an oil ( 8.89 kg ) GC analysis $42 \% R_{\mathrm{t}} 7.6 \mathrm{~min}$ (ester 6) and $51 \% R_{\mathrm{t}} 18.2 \mathrm{~min}$ (products 8 and 9). The oil was passed through a short-path distillation apparatus at $130^{\circ} \mathrm{C} 0.5 \mathrm{mbar}$ to recover the oxadiazole ester 6 and to give products 8 and 9 as a red oil ( $5.66 \mathrm{~kg}, 97 \%$ ). GC analysis $92 \% R_{t} 18.2 \mathrm{~min}$. A sample was chromatographed on silica with ethyl acetate-hexane (1:9) to give the pure products 8 and 9 as a colourless oil (Found: C, $59.25 ; \mathrm{H}, 7.5 ; \mathrm{N}, 11.45$. $\mathrm{C}_{18} \mathrm{H}_{27} \mathrm{~N}_{3} \mathrm{O}_{5}$ requires $\mathrm{C}, 59.15 ; \mathrm{H}, 7.45 ; \mathrm{N}, 11.5 \%$; $\delta_{\mathrm{H}}\left(\mathrm{CD}_{2} \mathrm{Cl}_{2}\right)$ 0.9-1.1 (4 H, m, cyclopropyl $\left.\mathrm{CH}_{2} \mathrm{CH}_{2}\right), 1.19(3 \mathrm{H}$, $2 \times \mathrm{t}, J 7.0, \mathrm{Me})$, 1.3-1.4 ( $9 \mathrm{H}, \mathrm{s}, \mathrm{CMe}_{3}$ ), $2.05(1 \mathrm{H}, \mathrm{m}$, cyclopropyl CH), $3.73(1 \mathrm{H}, \mathrm{d}, J 9.0, \mathrm{CHCO})$ and $4.13(2 \mathrm{H}$, $2 \times \mathrm{q}, J 7.0, \mathrm{OCH}_{2}$ ).
(2R, $3^{\prime} \mathrm{R}$ )- and (2S, $3^{\prime} \mathrm{R}$ )-2-(3-Cyclopropyl-1,2,4-oxadiazol-5-yl)-2-(N-tert-butoxycarbonylpyrrolidin-3-yl)ethanol 10 and 11.-Sodium borohydride $(2.57 \mathrm{~kg}, 67.9 \mathrm{~mol})$ was added to a solution of the esters 8 and $9(6.2 \mathrm{~kg}, 16 \mathrm{~mol})$ in tetrahydrofuran (THF) ( $18.6 \mathrm{dm}^{3}$ ) at $-5^{\circ} \mathrm{C}$. Methanol ( $13.8 \mathrm{dm}^{3}$ ) was added to the suspension over 1 h maintaining the temperature at $0^{\circ} \mathrm{C}$. The reaction mixture was stirred at $-5^{\circ} \mathrm{C}$ for 3 h , cooled to $-20^{\circ} \mathrm{C}$ and added to a mixture of ethyl acetate ( $30 \mathrm{dm}^{3}$ ) and hydrochloric acid ( $146 \mathrm{dm}^{3}, 2 \mathrm{~mol} \mathrm{dm}{ }^{-3}$ ) at $<-5^{\circ} \mathrm{C}$. The mixture was stirred for 15 min then the aqueous layer separated and extracted with ethyl acetate ( $5 \times 15 \mathrm{dm}^{3}$ ). The combined organic phases were washed sequentially with hydrochloric acid ( $13 \mathrm{dm}^{3}, 2 \mathrm{~mol} \mathrm{dm}{ }^{-3}$ ) and brine ( $3 \times 13 \mathrm{dm}^{3}$ ) and dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$. The solution was evaporated under reduced pressure to give alcohols 10 and 11 as an oil ( $4.65 \mathrm{~kg}, 85 \%$ ), GC/MS analysis $95 \%$ ratio of 10 and 11 (1.9:1). A sample was chromatographed on silica with diethyl ether--hexane (3:2) to give the pure less polar ( $2 R, 3 R^{\prime}$ ) enantiomer 10 as a solid, m.p. $71-74^{\circ} \mathrm{C}$ (Found: $\mathrm{C}, 59.3 ; \mathrm{H}, 7.75 ; \mathrm{N}, 12.8 . \mathrm{C}_{16} \mathrm{H}_{25} \mathrm{~N}_{3} \mathrm{O}_{4}$ requires $\mathrm{C}, 59.45 ; \mathrm{H}, 7.8 ; \mathrm{N}, 13.0 \%),[\alpha]_{\mathrm{D}}+1.7\left(c 1\right.$ in $\left.\mathrm{CH}_{2} \mathrm{Cl}_{2}\right)$; $\delta_{\mathrm{H}}\left[\left(\mathrm{CD}_{3}\right)_{2} \mathrm{SO}\right.$ at $\left.60^{\circ} \mathrm{C}\right] 0.90$ and $1.02(4 \mathrm{H}, \mathrm{m}$, cyclopropyl $\left.\mathrm{CH}_{2} \mathrm{CH}_{2}\right), 1.39\left(9 \mathrm{H}, \mathrm{s}, \mathrm{Me}_{3}\right), 1.50(1 \mathrm{H}, \mathrm{m}$, pyrrolidine 4-H), $1.75(1 \mathrm{H}, \mathrm{m}$, pyrrolidine $4-\mathrm{H}), 2.08(1 \mathrm{H}, \mathrm{m}$, cyclopropyl CH$)$, $2.50(1 \mathrm{H}, \mathrm{m}$, pyrrolidine $3-\mathrm{H})$, $3.14(2 \mathrm{H}, \mathrm{m}, \mathrm{CH}$ and pyrrolidine $5-\mathrm{H}$ ), $3.28(1 \mathrm{H}$, ddd, $J 2.8,9.0$ and 11.0 , pyrrolidine $5-\mathrm{H}), 2.98(1 \mathrm{H}$, dd, $J 10.5$ and 19.5 , pyrrolidine $2-\mathrm{H}$ ), $3.56(1 \mathrm{H}$,

[^0]dd, $J 7.5$ and 10.5 , pyrrolidine $2-\mathrm{H})$ and $3.71(2 \mathrm{H}, \mathrm{d}, J 6.0$, $\mathrm{CH}_{2} \mathrm{O}$ ); $\delta_{\mathrm{C}} 27.92\left(\mathrm{Me}_{3}\right), 28.52$ (pyrrolidine 4- $\mathrm{CH}_{2}$ ), 37.64 (pyrrolidine $3-\mathrm{CH}$ ), $43.76(\mathrm{CH}), 44.60$ (pyrrolidine $5-\mathrm{CH}_{2}$ ), 49.00 (pyrrolidine 2- $\mathrm{CH}_{2}$ ), $61.74\left(\mathrm{CH}_{2} \mathrm{O}\right.$ ), $77.94(\mathrm{C}-\mathrm{O}), 153.17$ $(\mathrm{C}=\mathrm{O}$ ), 171.23 (oxadiazole $\mathrm{C}-3$ ) and 179.31 (oxadiazole $\mathrm{C}-5$ ).
(2R,3'R)- and (2S, $3^{\prime} \mathrm{R}$ )-2-(3-Cyclopropyl-1,2,4-oxadiazol-5-yl)-2-(N-tert-butoxycarbonylpyrrolidin-3-yl)ethyl Methanesulfonate 12 and 13.-Triethylamine ( $3.97 \mathrm{dm}^{3}, 28.4 \mathrm{~mol}$ ) was added to a solution of the alcohols 10 and $11(4.6 \mathrm{~kg}, 14.2 \mathrm{~mol})$ in ethyl acetate ( $46 \mathrm{dm}^{3}$ ) cooled at $-20^{\circ} \mathrm{C}$. Methanesulfonyl chloride ( $1.32 \mathrm{dm}^{3}, 17.1 \mathrm{~mol}$ ) was added to the solution over 30 min maintaining the temperature at -20 to $-15^{\circ} \mathrm{C}$. The reaction mixture was stirred at $-20^{\circ} \mathrm{C}$ for 45 min and then quenched by slow addition of hydrochloric acid ( $14.1 \mathrm{dm}^{3}, 2$ $\mathrm{mol} \mathrm{dm}{ }^{-3}$ ). The aqueous layer was separated and extracted with ethyl acetate $\left(3 \times 3 \mathrm{dm}^{3}\right)$. The combined organic phases were washed with brine ( $4 \times 6.4 \mathrm{dm}^{3}$ ) and concentrated. Some solid which separated was removed by filtration and the filtrate evaporated under reduced pressure to give the methanesulfonates 12 and 13 as an oil $(5.8 \mathrm{~kg}, 100 \%)$. A sample was chromatographed on silica with diethyl ether-hexane (9:4) to give less polar ( $2 R, 3^{\prime} R$ ) enantiomer 12 as a solid, m.p. $85-88^{\circ} \mathrm{C}$ (Found: C, $50.9 ; \mathrm{H}, 6.8 ; \mathrm{N}, 10.4 ; \mathrm{S}, 8.0 . \mathrm{C}_{17} \mathrm{H}_{27} \mathrm{~N}_{3} \mathrm{O}_{6} \mathrm{~S}$ requires C, $50.85 ; \mathrm{H}, 6.8 ; \mathrm{N}, 10.45 ; \mathrm{S}, 8.0 \%$ ) ; $\alpha]_{\mathrm{D}}-25.3\left(c \mathrm{l}\right.$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ ); $\delta_{\mathrm{H}}\left(\mathrm{CD}_{2} \mathrm{Cl}_{2}\right) 0.85-1.05\left(4 \mathrm{H}, \mathrm{m}\right.$, cyclopropyl $\left.\mathrm{CH}_{2} \mathrm{CH}_{2}\right), 1.35(9$ H , br s, $\mathrm{CMe}_{3}$ ), $1.51(1 \mathrm{H}, \mathrm{m}$, pyrrolidine $4-\mathrm{H}), 1.79(1 \mathrm{H}, \mathrm{m}$, pyrrolidine $4-\mathrm{H}$ ), 2.01 ( $1 \mathrm{H}, \mathrm{dt}, J 5.0$ and 8.5 cyclopropyl CH ), $2.47\left(1 \mathrm{H}, \mathrm{m}\right.$, pyrrolidine $3-\mathrm{H}$ ), $2.89\left(3 \mathrm{H}, \mathrm{s}, \mathrm{MeSO}_{2}\right), 2.98-3.31$ ( $4 \mathrm{H}, \mathrm{m}$, pyrrolidine $2-\mathrm{H}_{2}$ and $5-\mathrm{H}_{2}$ ), $3.58(1 \mathrm{H}, \mathrm{q}, J 9.5, \mathrm{CH}$ ) and 4.38 and $4.4\left(2 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{2} \mathrm{O}\right) ; \delta_{\mathrm{C}}\left[\left(\mathrm{CD}_{3}\right)_{2} \mathrm{SO}\right.$ at $\left.60^{\circ} \mathrm{C}\right] 27.87$ $\left(\mathrm{Me}_{3}\right), 28.22$ (pyrrolidine 4- $\mathrm{CH}_{2}$ ), $36.62\left(\mathrm{MeSO}_{2}\right), 37.64$ (pyrrolidine $3-\mathrm{CH}$ ), $40.10\left(\mathrm{CH}\right.$ ), 44.57 (pyrrolidine $5-\mathrm{CH}_{2}$ ), 48.58 (pyrrolidine 2- $\mathrm{CH}_{2}$ ), $68.76\left(\mathrm{CH}_{2} \mathrm{O}\right), 78.06(\mathrm{C}-\mathrm{O}), 153.11$ $(\mathrm{C}=\mathrm{O}$ ), 171.56 (oxadiazole $\mathrm{C}-3$ ) and 177.15 (oxadiazole $\mathrm{C}-5$ ).
(3R,4R)-3-(3-Cyclopropyl-1,2,4-oxadiazol-5-yl)-1-azabicyclo[2.2.1]heptane 14.-(a) From methanesulfonates 12 and 13. Trifluoroacetic acid ( $12.0 \mathrm{dm}^{3}$ ) was added dropwise over 1 h to a solution of the methanesulfonates 12 and $13(5.9 \mathrm{~kg}, 14.7 \mathrm{~mol})$ in tert-butyl alcohol ( $3.0 \mathrm{dm}^{3}$ ) at $20-25^{\circ} \mathrm{C}$. The reaction mixture was stirred at $20^{\circ} \mathrm{C}$ for 1 h , diluted with tert-butyl alcohol ( $24 \mathrm{dm}^{3}$ ), and aqueous sodium carbonate ( $80 \mathrm{dm}^{3}, 10 \%$ ) was added to adjust the pH to 7.5 . The mixture was heated to $40^{\circ} \mathrm{C}$ for 1.5 h maintaining the pH at 7.5 by addition of sodium carbonate solution. Heating at $40^{\circ} \mathrm{C}$ was continued for a further 1.5 h maintaining the pH at 8.5 . The pH was finally increased to 9.5 and the reaction mixture extracted with toluene $\left(20 \mathrm{dm}^{3}\right.$ and $4 \times 10 \mathrm{dm}^{3}$ ). The combined organic phases were dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$ and evaporated under reduced pressure to give the bicyclo[2.2.1]heptanes 14 and 15 as an oil ( 2.08 kg ). GC analysis $65 \% R_{\mathrm{t}} 11.0 \mathrm{~min}$ (product 14) and $28 \% R_{\mathrm{t}} 10.6 \mathrm{~min}$ (product 15). The crude product was chromatographed on silica with ethyl acetate-methanol $(5: 1)$ and then methanol to give the pure enantiomer 14 as a crystalline solid ( $803 \mathrm{~g}, 27 \%$ ), m.p. $60-61^{\circ} \mathrm{C}$ (Found: C, 64.35; $\mathrm{H}, 7.4 ; \mathrm{N}, 20.45 . \mathrm{C}_{11} \mathrm{H}_{15} \mathrm{~N}_{3} \mathrm{O}$ requires $\mathrm{C}, 64.35 ; \mathrm{H}, 7.35 ; \mathrm{N}, 20.45 \%) ;[\alpha]_{\mathrm{D}}-1.7\left(c 1\right.$ in $\left.\mathrm{H}_{2} \mathrm{O}\right)$; $\delta_{\mathrm{H}}\left(\mathrm{CD}_{2} \mathrm{Cl}_{2}\right)$ 0.9-1.1 (4 H, m, cyclopropyl $\left.\mathrm{CH}_{2} \mathrm{CH}_{2}\right), 1.25(1 \mathrm{H}$, $\mathrm{m}, 5-\mathrm{H}), 1.67(1 \mathrm{H}, \mathrm{dt}, J 4.5$ and $11.2,5-\mathrm{H}), 2.02(1 \mathrm{H}, \mathrm{dt}, J 5.5$ and 8.0 , cyclopropyl CH ), 2.35 ( 1 H , d quin, $J 1.0$ and $10.0,7-$ H), $2.49(1 \mathrm{H}$, dddd, $J 2.2,5.0,10.5$ and 13.0, exo $6-\mathrm{H}), 2.69$ (1 H , ddd, $J 1.0,2.0$ and $10.0,7-\mathrm{H}), 2.7-2.85(3 \mathrm{H}, \mathrm{m}, 3-\mathrm{H}, 4-\mathrm{H}$ and endo $6-\mathrm{H}), 2.90(1 \mathrm{H}$, ddd, $J 2.5,8.0$ and $12.0,2-\mathrm{H})$ and $3.02(1$ H , ddd, $J 2.5,7.8$ and $12.0,2-\mathrm{H}) ; \delta_{\mathrm{C}}\left(\mathrm{CD}_{2} \mathrm{Cl}_{2}\right) 7.73$ (cyclopropyl CH ), 8.4 (cyclopropyl $\mathrm{CH}_{2}$ ), $31.07\left(5-\mathrm{CH}_{2}\right), 41.35(4-\mathrm{CH})$, $43.77(3-\mathrm{CH}), 54.76\left(6-\mathrm{CH}_{2}\right), 59.68\left(7-\mathrm{CH}_{2}\right), 61.39\left(2-\mathrm{CH}_{2}\right)$, 173.03 (oxadiazole $\mathrm{C}-3$ ) and 182.20 (oxadiazole C-5); Chiral HPLC analysis (Bakerbond Chiral AGP) $>98.5 \%$ e.e. Latter
fractions from the column gave an exo-endo mixture of products 14 and 15 ( $406 \mathrm{~g}, 14 \%$ ) GC analysis $17 \% R_{\mathrm{t}} 11.0 \mathrm{~min}$ (product 14) and $83 \% R_{\mathrm{t}} 10.6 \mathrm{~min}$ (product 15).
(b) From methanesulfonate 12. The methanesulfonate 12 (200 $\mathrm{mg}, 0.5 \mathrm{mmol}$ ) in tert-butyl alcohol ( $5 \mathrm{~cm}^{3}$ ) was treated as above to give the crude enantiomer $14(60 \mathrm{mg}, 60 \%)$ GC analysis single peak $R_{1} 11.0 \mathrm{~min}$.
(c) By epimerisation of endo-enantiomer 15. A solution of the endo enantiomer enriched mixture of products 14 and $15(120 \mathrm{~g}$, ratio $17: 83$ ) and potassium tert-butoxide ( 7.5 g ) in ethanol ( 600 $\mathrm{cm}^{3}$ ) was heated under reflux for 30 min . GC analysis $76 \% R_{\mathrm{t}}$ 11.0 min (product 14) and $24 \% R_{\mathrm{t}} 10.6 \mathrm{~min}$ (product 15). The solution was evaporated and the residue partitioned between ethyl acetate ( $400 \mathrm{~cm}^{3}$ ) and water ( $50 \mathrm{~cm}^{3}$ ). The aqueous phase was separated and extracted with ethyl acetate ( $3 \times 50 \mathrm{~cm}^{3}$ ). The combined organic phases were dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$ and evaporated to an oil ( $118 \mathbf{g}$ ). A solution of toluene- $p$-sulfonic acid monohydrate ( $109 \mathrm{~g}, 0.57 \mathrm{~mol}$ ) in ethyl acetate ( $500 \mathrm{~cm}^{3}$ ) was distilled to give a solution ( $350 \mathrm{~cm}^{3}$ ) of the anhydrous acid, which was diluted with isopropyl alcohol ( $120 \mathrm{~cm}^{3}$ ). The exoenantiomer enriched oil, in a mixture of ethyl acetate ( $260 \mathrm{~cm}^{3}$ ) and isopropyl alcohol ( $116 \mathrm{~cm}^{3}$ ), was added to the acid solution and the solution stirred at $20^{\circ} \mathrm{C}$ for 2 h to crystallise. The mixture was cooled to $5^{\circ} \mathrm{C}$ for 1 h and the solid collected by filtration. GC analysis $98.4 \% R_{\mathrm{t}} 11.0 \mathrm{~min}$ (product 14). The solid was suspended in boiling ethyl acetate ( $1.2 \mathrm{dm}^{3}$ ) for 2 h and then cooled to $20^{\circ} \mathrm{C}$. The solid was collected to give the enantiomer 14 as the crystalline toluene-p-sulfonate salt ( 117 g , $51 \%$ ), m.p. 132-133 ${ }^{\circ} \mathrm{C}$ (Found: C, 57.0; H, 6.15; N, 11.2; S, 8.5. $\mathrm{C}_{11} \mathrm{~N}_{15} \mathrm{~N}_{3} \mathrm{O} \cdot \mathrm{C}_{7} \mathrm{H}_{8} \mathrm{O}_{3} \mathrm{~S}$ requires $\mathrm{C}, 57.3 ; \mathrm{H}, 6.15 ; \mathrm{N}, 11.15$; S , $8.5 \%$ ); $[\alpha]_{\mathrm{D}}-2.4\left(c 1\right.$ in $\mathrm{H}_{2} \mathrm{O}$ ); GC analysis $99.5 \% R_{\mathrm{t}} 11.0$ min ( $3 R, 4 R$ )-enantiomer.
(3S,4R)-3-(3-Cyclopropyl-1,2,4-oxadiazol-5-yl)-1-azabicyclo[2.2.1]heptane 15.-A sample of the exo-endo mixture of products 14 and 15, obtained from the column in the preparation of the exo enantiomer 14, was chromatographed on silica with ethyl acetate-methanol ( $5: 1$ ) and then methanol to give the endo enantiomer 15 as an oil. The oil ( $1.1 \mathrm{~g}, 5.4 \mathrm{mmol}$ ) was dissolved in ethyl acetate ( $20 \mathrm{~cm}^{3}$ ) and treated with a solution of toluene-p-sulfonic acid monohydrate $(1.0 \mathrm{~g}, 5.3$ mmol ) in ethyl acetate ( $30 \mathrm{~cm}^{3}$ ) at $20^{\circ} \mathrm{C}$. The crystalline solid was collected to give the enantiomer 15 as the toluene- $p$ -
sulfonate salt $\left(1.8 \mathrm{~g}, 89 \%\right.$ ), m.p. $138-139^{\circ} \mathrm{C}$ (Found: $\mathrm{C}, 57.3 ; \mathrm{H}$, $6.15 ; \mathrm{N}, 11.35 ; \mathrm{S}, 8.5 . \mathrm{C}_{11} \mathrm{H}_{15} \mathrm{~N}_{3} \mathrm{O} \cdot \mathrm{C}_{7} \mathrm{H}_{8} \mathrm{O}_{3} \mathrm{~S}$ requires $\mathrm{C}, 57.3$; $\mathrm{H}, 6.15 ; \mathrm{N}, 11.15 ; \mathrm{S}, 8.5 \%) ;[\alpha]_{\mathrm{D}}+35.8\left(c 1\right.$ in $\left.\mathrm{H}_{2} \mathrm{O}\right) ; \delta_{\mathrm{H}}($ free base in $\mathrm{CD}_{2} \mathrm{Cl}_{2}$ ) $0.95-1.05\left(4 \mathrm{H}, \mathrm{m}\right.$, cyclopropyl $\left.\mathrm{CH}_{2} \mathrm{CH}_{2}\right), 1.11$ $(1 \mathrm{H}, \mathrm{m}, 5-\mathrm{H}), 1.41(1 \mathrm{H}, \mathrm{m}, 5-\mathrm{H}), 2.03(1 \mathrm{H}, \mathrm{dt}, J 5.0$ and 8.0 , cyclopropyl CH), 2.45-2.65 ( $3 \mathrm{H}, \mathrm{m}, 6-\mathrm{H}$ and $7-\mathrm{H}_{2}$ ), 2.7-3.0 ( 3 $\mathrm{H}, \mathrm{m}, 3-\mathrm{H}, 4-\mathrm{H}$ and $6-\mathrm{H}), 3.15(1 \mathrm{H}$, ddd, $J 2.5,11.0$ and $12.5,2-$ H ) and $3.31(1 \mathrm{H}$, dddd, $J 2.0,5.0,9.0$ and $11.0,2-\mathrm{H})$; $\delta_{\mathrm{C}}($ free base in $\mathrm{CD}_{2} \mathrm{Cl}_{2}$ ) 7.71 (cyclopropyl CH ), 8.42 (cyclopropyl $\mathrm{CH}_{2}$ ), $26.02\left(5-\mathrm{CH}_{2}\right), 40.44(4-\mathrm{CH}), 43.36(3-\mathrm{CH}), 55.17(6-$ $\mathrm{CH}_{2}$ ), $59.33\left(7-\mathrm{CH}_{2}\right), 62.83\left(2-\mathrm{CH}_{2}\right), 173.08$ (oxadiazole C-3) and 181.29 (oxadiazole C-5); GC analysis $99.0 \% R_{\mathrm{t}} 10.6 \mathrm{~min}$ $(3 S, 4 R)$-enantiomer.

## Acknowledgements

We thank P. V. Byway for GC/mass spectral data.

## References

1 J. Saunders, A. M. MacLeod, K. J. Merchant, G. A. Showell, R. J. Snow, L. J. Street and R. Baker, J. Chem. Soc., Chem. Commun., 1988, 1618; L. J. Street, R. Baker, T. Book, C. O. Kneen, A. M. MacLeod, K. J. Merchant, G. A. Showell, J. Saunders, R. H. Herbert, S. B. Freedman and E. A. Harley, J. Med. Chem., 1990, 33, 2690; G. A. Showell, R. Baker, J. Davis, R. Hargreaves, S. B. Freedman, K. Hoogsteen, S. Patel and R. J. Snow, J. Med. Chem., 1992, 35, 911.
2 B. S. Orlek, F. E. Blaney, F. Brown, M. S. G. Clark, M. S. Hadley, J. Hatcher, G. J. Riley, H. E. Rosenberg, H. J. Wadsworth and P. Whyman, J. Med. Chem., 1991, 34, 2726.
3 R. Baker, G. A. Showell, L. J. Street, J. Saunders, K. Hoogsteen, S. B. Freedman and R. J. Hargreaves, J. Chem. Soc., Chem. Commun., 1992, 817.
4 I. F. Cottrell, D. Hands, D. J. Kennedy, K. J. Paul, S. H. B. Wright and K. Hoogsteen, J. Chem. Soc., Perkin Trans. 1, 1991, 1091.
5 M. Hashimoto, Y. Eda, Y. Osanai, T. Iwai and S. Aoki, Chem. Lett., 1986, 893.
6 K. Soai, H. Oyamada and A. Ookawa, Synth. Commun., 1982, 12, 463.

7 W. J. Fanshawe, V. J. Bauer, S. R. Safir, D. A. Blickens and S. J. Riggi, J. Med. Chem., 1969, 12, 381.

Paper 3/01347B
Received 8th March 1993
Accepted 1st April 1993


[^0]:    * $1 \mathrm{bar}=10^{5} \mathrm{~Pa}$

